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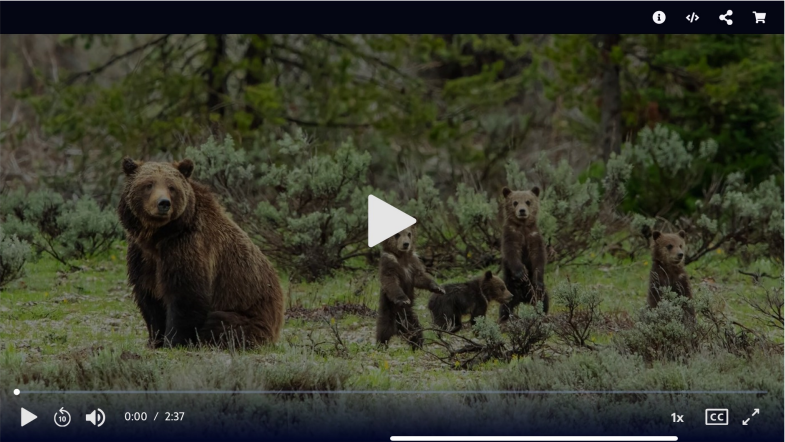
BIG CATS 24/7NEW SEASON STREAM NOW PBS

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S42E15: GRIZZLY 399: QUEEN OF THE TETONS | CLIP

Meet the Most Famous Grizzly in the World



STREAM WITH PASSPORT



Grizzly 399: Queen of the Tetons

PROVIDING SUPPORT FOR PBS.ORG

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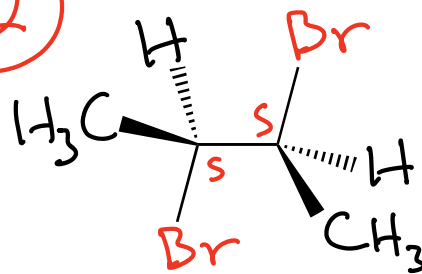
BIG CATS 24/7NEW SEASON STREAM NOW PBS

Only carbocations rearrange!
Halonium ions (3-membered rings) DO NOT

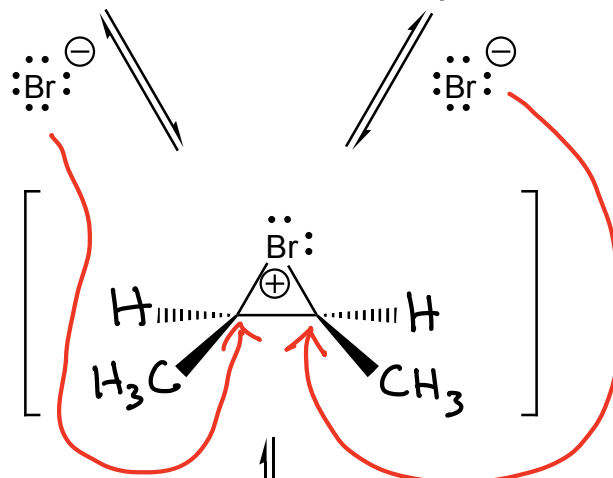
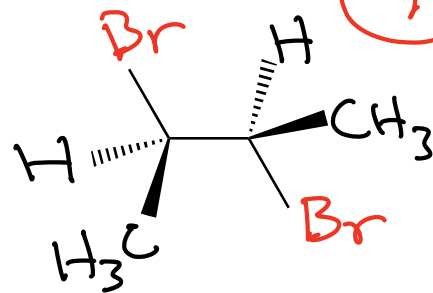


To get stereochemistry correct when predicting reaction products, you need to analyze each reaction on a case by case basis

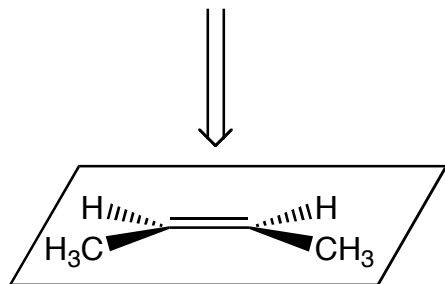
2



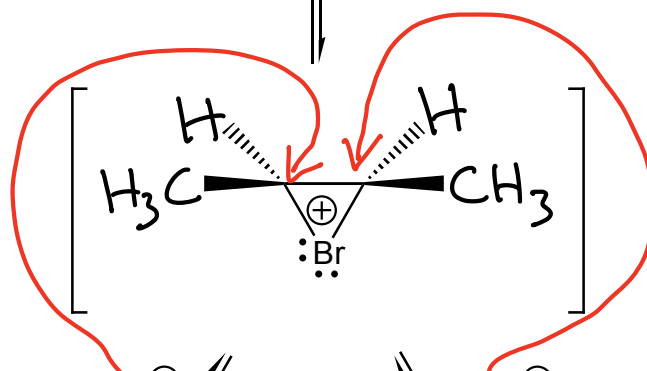
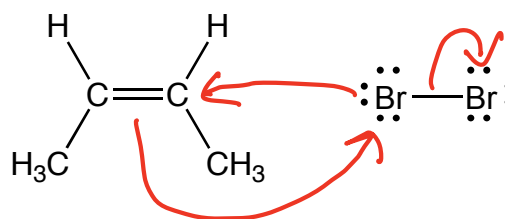
1



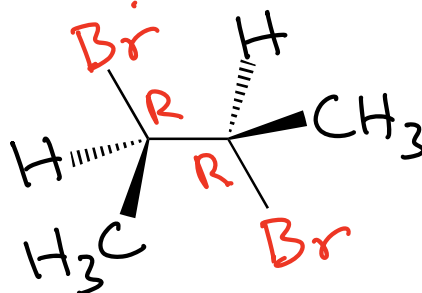
Attack from the top face



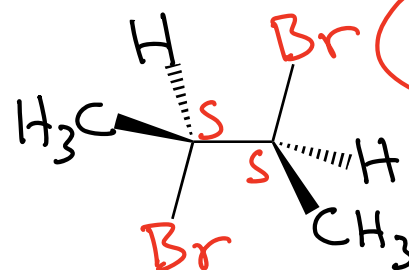
Attack from the bottom face

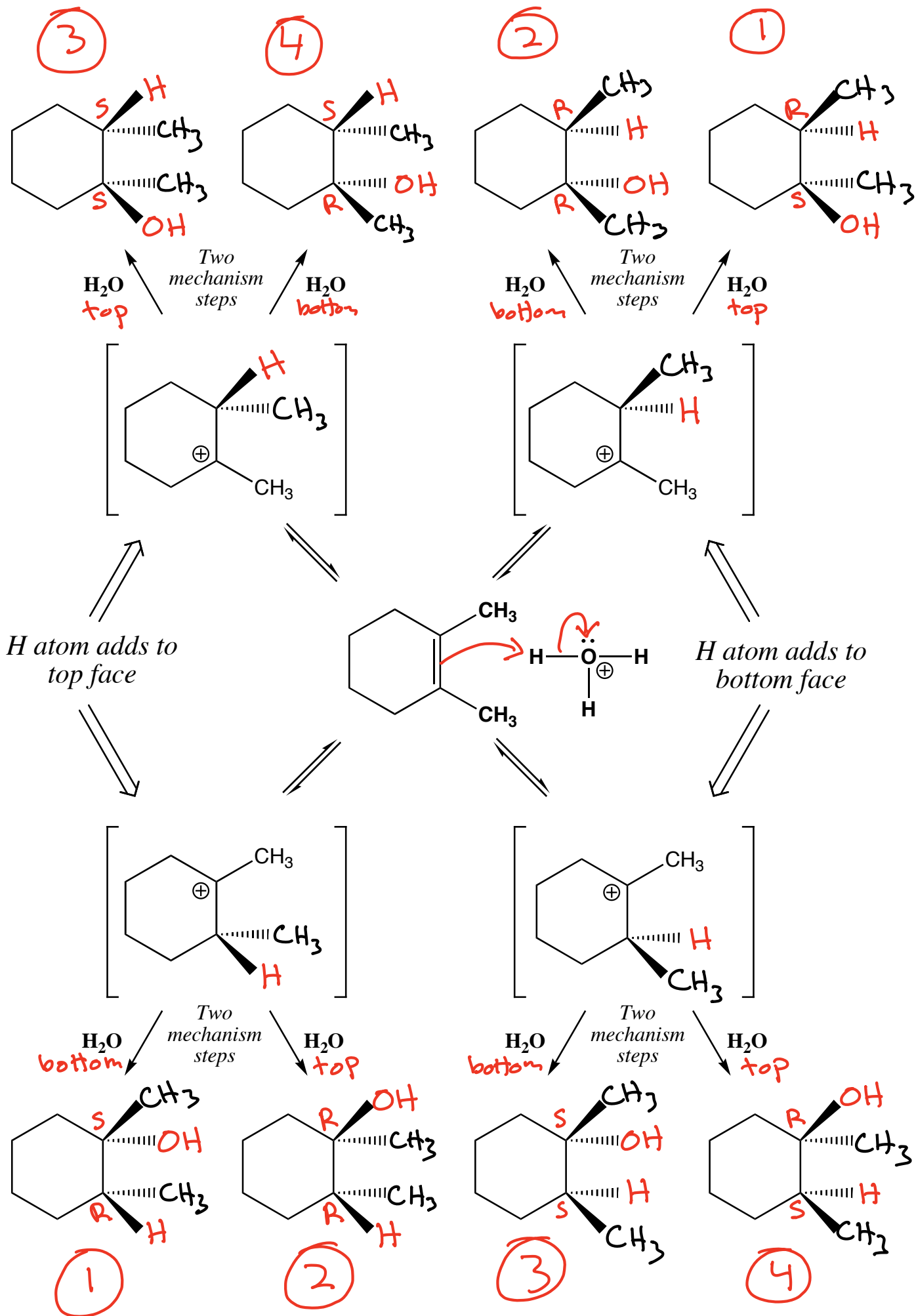


1

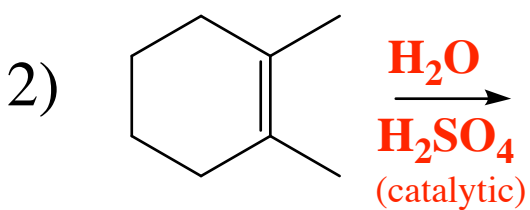
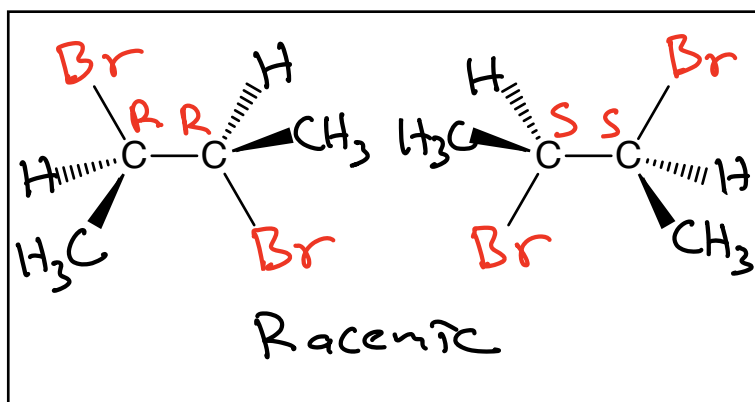
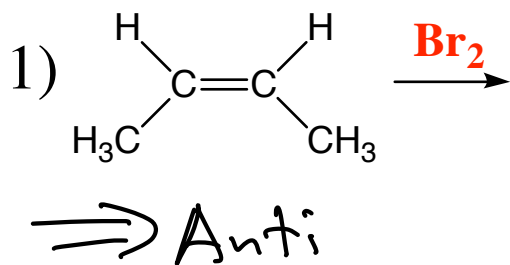


2

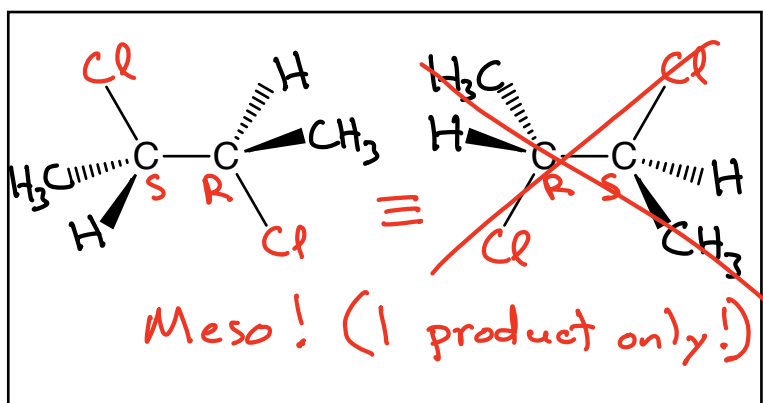
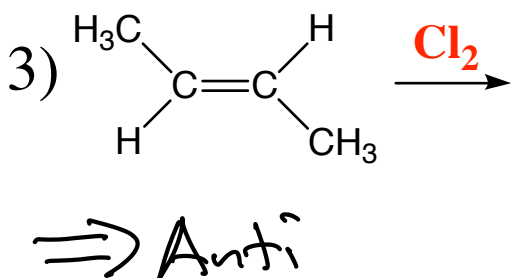
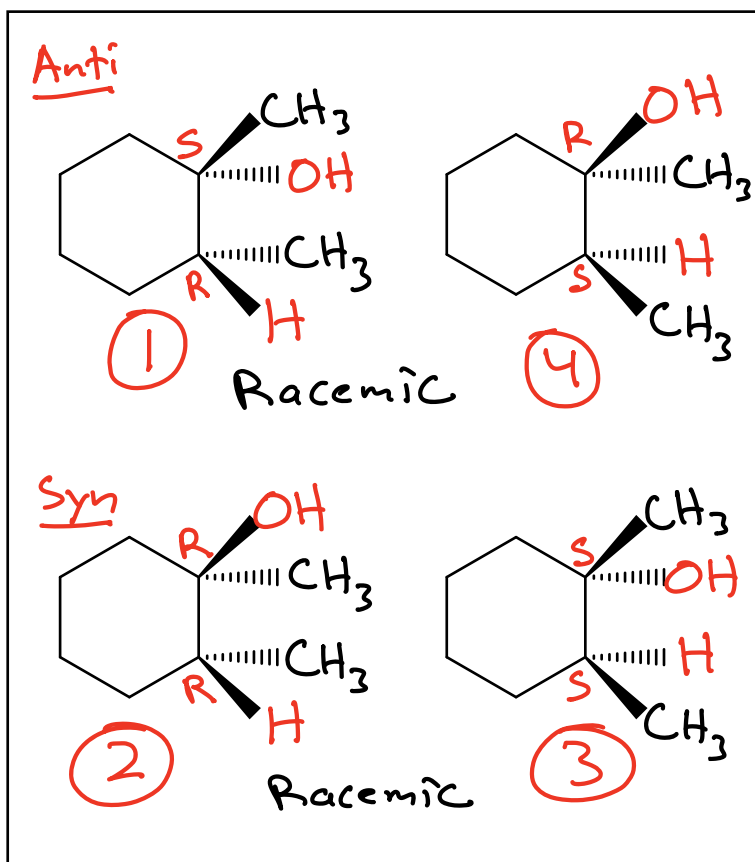




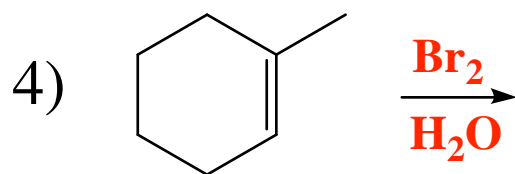
Examples



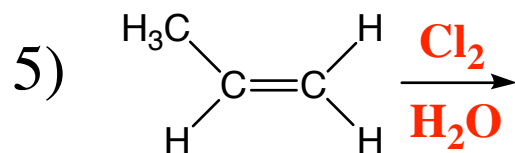
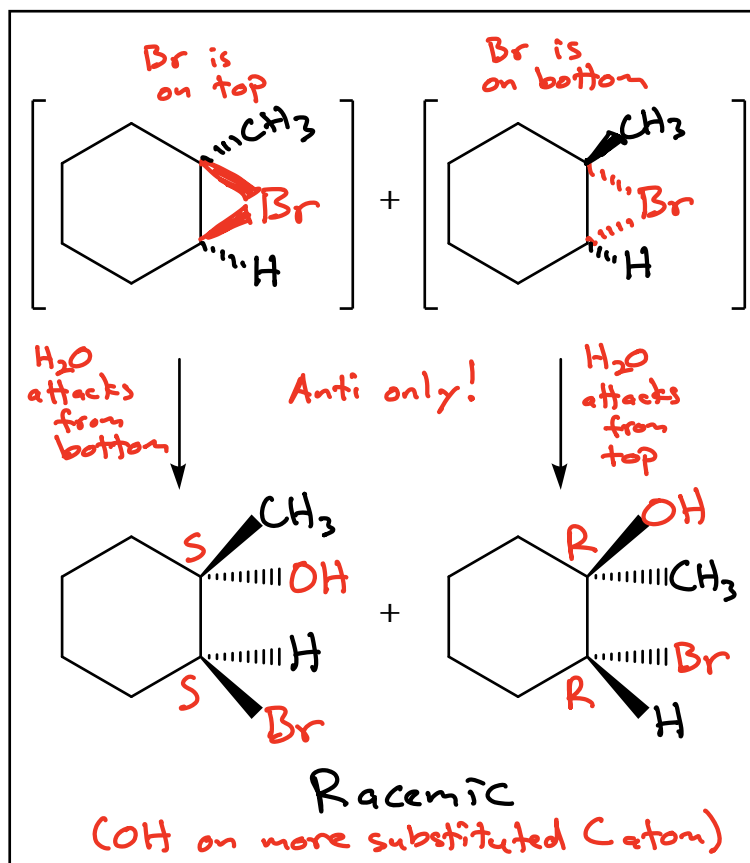
$\Rightarrow$ Mixed $\rightarrow$ Anti and Syn
 $\Rightarrow$ Markovnikov



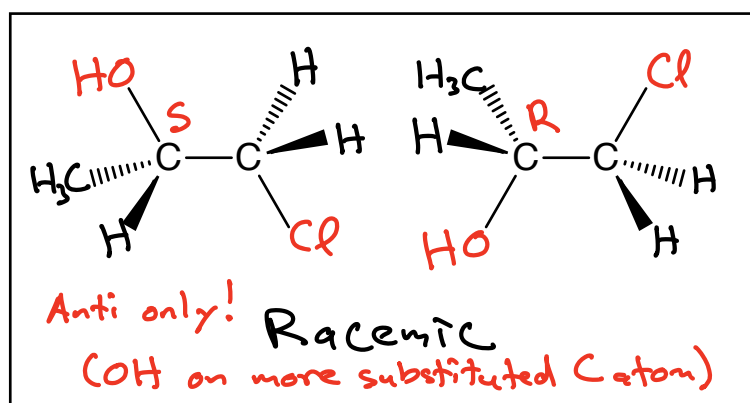
More Examples



⇒ Anti
⇒ Markovnikov



⇒ Anti
⇒ Markovnikov



Who do you call when you need help?

Nurse

[nɜːs] noun

lifesaving superhero, patient,
smile bringing, kind, lives to heal.
Kind of a big deal.

nurse

[nɜːs] *noun*

the first person you see after saying, "hold my beer and watch this!"

When studying OChem $\rightarrow$ Call a NIRRS
Learn each of these things for every
reaction $\rightarrow$ then you will be able to
predict mechanisms and therefore products

Nature of the reaction; what is the starting material/product? (i.e. alkene converted to an alcohol)

Intermediate (or "Important transition state" if applicable) of the reaction, the key to the mechanism (carbocation, halonium ion, etc.)

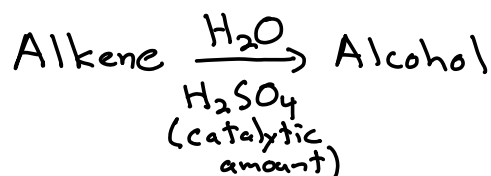
Reagents Learn the exact way to designate the reagents for each reaction

Regiochemistry What is the regiochemistry of addition? (Markovnikov, non-Markovnikov, etc.)

Stereochemistry of addition (anti, syn or mixed)



Carbocation
Markovnikov
Mixed



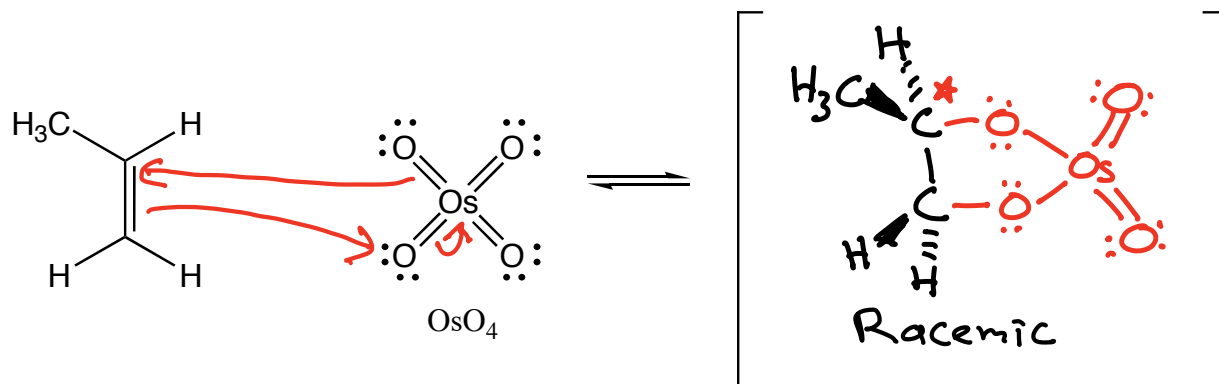
Carbocation
Markovnikov
Mixed

Ozzy Osbourne

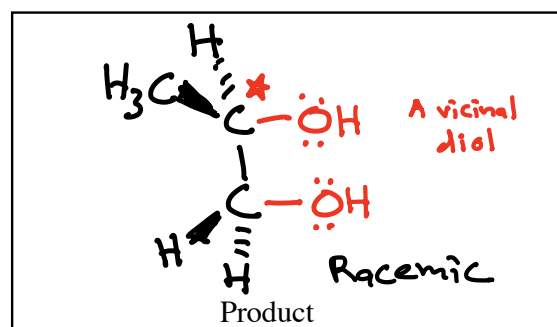


Ozzy Osborne Reaction → "Sin" (Syn)

OsO_4 Partial Mechanism



2. $\text{NaHSO}_3 / \text{H}_2\text{O}$
(Chemist opens up flask) $\nparallel$ Not responsible for mechanism



Summary:

The mechanism involves a cyclic osmate ester, explaining the syn (sin!) stereochemistry of addition.

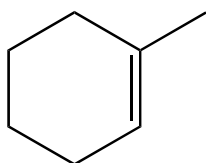
Regiochemistry:

N/A

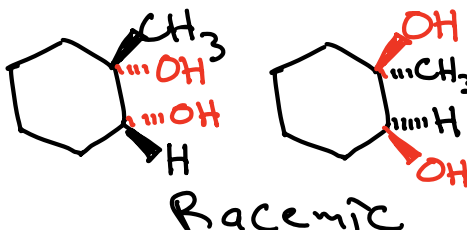
Stereochemistry:

Syn (Sin!)

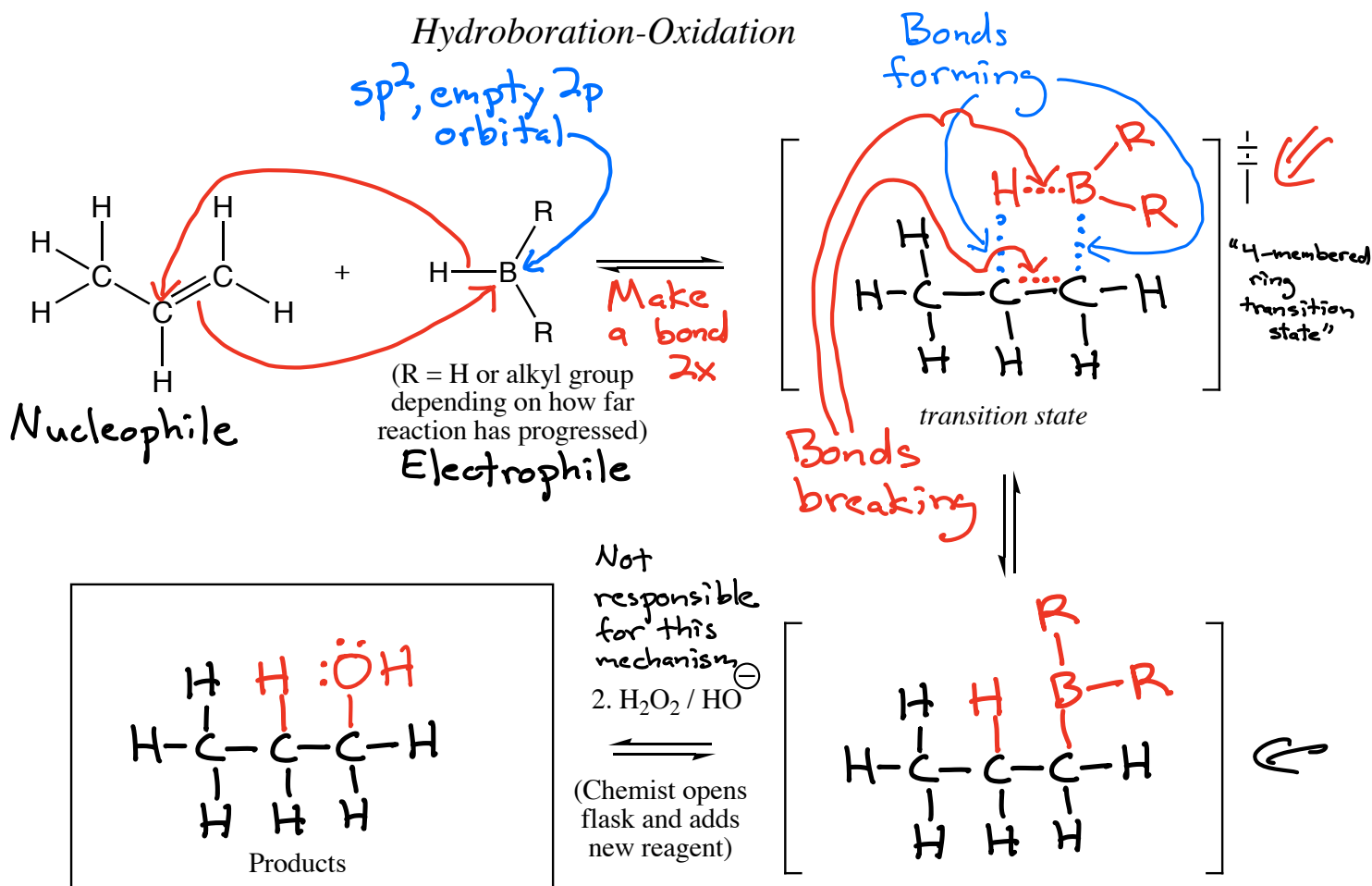
Example:



1. OsO_4
2. $\text{NaHSO}_3 / \text{H}_2\text{O}$



Hydroboration-Oxidation



$\text{H} \rightarrow$ More substituted C atom
 $\text{OH} \rightarrow$ Less substituted C atom

Steric strain in the first transition state

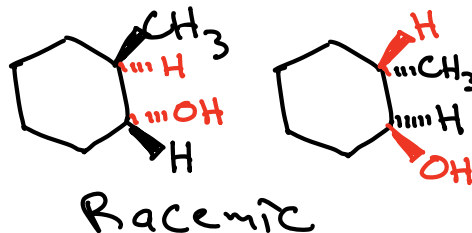
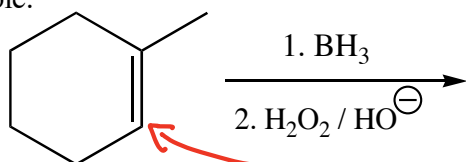
H and OH are syn

Summary: The π bond of the alkene attacks the Lewis acid (electrophile) B atom at the same time a new bond forms between C and H. In 2nd step OH replaces $\text{B}(\text{R})_2$. "4-membered ring transition state"

Regiochemistry: Non-Markovnikov

Stereochemistry: Syn

Example:



Less substituted C atom